Copper-Catalyzed Synthesis of Trisubstituted Isoxazoles via a Cascade Cyclization−Migration Process

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S Supporting Information

[AB](#page-6-0)STRACT: [An atom-econ](#page-6-0)omical, catalytic, and regioselective synthesis of 3,4,5-trisubstituted isoxazoles has been successfully developed. Treatment of O-arylmethyl alkynyl oxime ethers with 5 mol % of $Cu(OTf)$, in chlorobenzene at reflux gave 4arylmethylisoxazoles in good to excellent yields via the sequential intramolecular addition of the oxime moiety to the alkyne with subsequent 1,3-migration of the arylmethyl group.

Heterocycles are well-known for their wide range of \blacktriangle biological properties.¹ Of the various bioactive heterocycles, isoxazoles have attracted considerable interest because of their wide reaching applic[at](#page-6-0)ions in medicinal chemistry² and material science.³ For this reason, considerable research efforts have been focused on the development of novel and e[ffi](#page-6-0)cient methods for the [s](#page-6-0)ynthesis of isoxazoles 3. 4

Although isoxazoles are generally synthesized according to an intermolecular [3 + 2] cycloaddition r[e](#page-6-0)action between an alkyne and a nitrile $oxide$, the direct construction of trisubstituted isoxazoles via a cycloaddition reaction with an internal alkyne has not been reported as frequently in the literature.^{6,7} Furthermore, these methods usually require harsh conditions and result in poor chemo- and regioselectivities. An alternativ[e s](#page-6-0)tepwise approach has also been reported for the selective and efficient construction of trisubstituted isoxazoles. This approach involves the preformation of an isoxazole ring bearing a functional group at the 4-position, such as a halogen, 8 silicon,⁹ boronic ester¹⁰ or aluminum,¹¹ which can be subsequently cross-coupled with an appropriate couplin[g](#page-6-0) partner to provide the [de](#page-6-0)sired substitute[d p](#page-6-0)roduct. Using a more straightforward approach, Chen et al.¹² and Campagne et al.¹³ independently reported the one-pot synthesis of trisubstituted isoxazoles using a domi[no](#page-6-0) sequence. This in[vol](#page-6-0)ved the construction of the isoxazole ring system through a palladium-catalyzed annulation of the corresponding acyclic precursor with subsequent cross-coupling of the resulting 4 isoxazolylpalladium intermediate.¹⁴ However, these reactions were not atom-economical, as the loss of a functional group would result in generation of ch[em](#page-6-0)ical waste in the course of the transformation.

From the perspective of atom economy alone, the direct synthesis of trisubstituted isoxazoles with high levels of chemoand regioselectivity in a one-pot sequence is both highly desirable and challenging. Herein, we report the direct synthesis of trisubstituted isoxazoles using a domino process involving the transition metal catalyzed cyclization of an O-benzyl alkynyl oxime ether with subsequent 1,3-migration of the benzyl groups.

The π-acidic transition-metal-catalyzed intramolecular addition of a heteroatom to an alkyne with subsequent migration of the substituent initially attached to the heteroatom is one of the most powerful strategies for the synthesis of heterocyclic compounds.¹⁵ Although the use of these transformations has effectively provided well-precedented access to benzofurans,¹⁶ indoles,¹⁷ b[en](#page-6-0)zothiophenes,¹⁸ furans,¹⁹ pyrans,²⁰ and pyrrolidine, 21 less is known about the synthesis of isoxazoles using t[his](#page-6-0) metho[dol](#page-6-0)ogy.²² Our wor[kin](#page-6-0)g hyp[oth](#page-6-0)esis f[or](#page-7-0) the current rese[arc](#page-7-0)h is outlined in Scheme 1. We recently reported the

Scheme 1. Synthesis of Di- and Trisubstituted Isoxazoles

development of a silver-catalyzed synthesis of disubstituted isoxazoles via the generation of a vinyl metal intermediate A which was subsequently protonated under acidic conditions to give the desired disubstituted isoxazole $2.^{23}$ It was envisaged that the benzyl group could migrate from the oxygen of the oxonium ion in A to the 4-position of [th](#page-7-0)e isoxazole core, leading to the formation of trisubstituted isoxazoles under suitable conditions.

Alkynyl oxime ether 1a bearing a p-methoxybenzyl group on the oxygen atom was selected as a model system for our initial screening work because it would generate a highly stabilized carbocation and enhance the migratory aptitude of the benzyl group. A series of alkynophilic catalysts were screened for their

Received: July 8, 2012 Published: September 24, 2012 ability to catalyze the synthesis of the trisubstituted isoxazole 3a (Table 1). When 1a was treated with $AuCl₃$ in dichloroethane

at reflux, the desired trisubstituted isoxazole 3a was obtained, albeit in a low yield (Table 1, entries 1 and 2).²⁴ A survey of other catalysts revealed $Cu(OTf)_2$ as the most effective catalyst (Table 1, entries 3−5). This reaction was found [to](#page-7-0) be strongly temperature dependent (Table 1, entries 6−10). Solvent polarity was also a significant factor for the efficiency of the reaction (Table 1, entries 8 and 9). It should be noted that nbutyl p-methoxybenzyl ether was obtained as a byproduct when the reaction was carried out in n -BuOH (Table 1, entry 10). This result indicated that the reaction proceeded via generation of p-methoxybenzyl cation. Following the screening process, the use of chlorobenzene at reflux was found to be optimal (Table 1, entries 6−9).

The general scope of the reaction was then examined under the optimized conditions (Table 1, entry 9). A variety of different O-p-methoxybenzyl alkynyl oxime ethers 1b−i were treated with $Cu(OTf)$ ₂ in refluxing chlorobenzene, affording

the corresponding trisubstituted isoxazole 3b−i in good to high yields (Table 2). Electron-rich and electron-poor aryl groups and an alkyl group were well tolerated at the $R¹$ position, affording the corresponding trisubstituted isoxazoles 3b−d (Table 2, entries 1−3) in good yields. Furthermore, the ester moiety on oxime ether 1e was compatible with the reaction conditions, providing isoxazole carboxylate 3e (Table 2, entry 4) in good yield and demonstrating the mild nature of the current transformation. The substituent at the R^2 position on the triple bond terminus could also be varied without any discernible issue under the optimized reaction conditions. For example, substrates 1f and 1g bearing aliphatic substituents performed well under the current reaction conditions providing the desired trisubstituted isoxazoles in high yields (Table 2, entries 5 and 6). It is noteworthy that the acidic proton of the terminal alkyne in oxime 1h did not have an adverse impact on the transformation and the desired product was obtained in high chemical yield (Table 2, entry 7). Unfortunately, when the silyl-protected substrate 1i was subjected to the current reaction conditions, partial desilylation was observed. Consequently, upon completion of the copper-catalyzed cyclization, the crude product was treated with TBAF to afford 4-(hydroxymethyl) isoxazole 3i in 65% yield (Table 2, entry 8). Additionally, this method for the synthesis of trisubstituted isoxazoles can be applied to a gram scale synthesis; thus, the copper-catalyzed reaction of 1 \mathbf{b} (1 g) afforded 3 \mathbf{b} in 63% yield (entry 9).

We then proceeded to investigate the scope for changing the substituents on the migrating group (Table 3). The p silyloxybenzyl group in oxime ether 4a successfully migrated to give the corresponding trisubstituted isoxazole 5a in high yield (Table 3, entry 1). Remarkably, the intro[du](#page-2-0)ction of a dimethylamino group in oxime ether 4b enhanced the efficiency of [th](#page-2-0)e domino reaction, which proceeded even at a lower temperature in refluxing dichloroethane (Table 3, entry 2). The substrate 4c bearing an o-methoxy group gave 5c in relatively lower yield, likely because of steric repulsion ([T](#page-2-0)able 3, entry 3). The introduction of an electron-withdrawing group such as an ester moiety completely inhibited the cyclizati[on](#page-2-0) reaction and only the oxime ether 4d was recovered (Table 3, entry 4). This result indicated that the rate-determining step involved the generation of the benzyl cation by C−O bo[nd](#page-2-0)

 a Isolated yields. b Reaction was carried out with Cu(OTf)₂ (5 mol %) followed by treatment with TBAF (1 equiv). ^cReaction was carried out with 1 g of 1b.

Table 3. Substitution Effect of the Migrating Group

^aReaction was carried out in $(\text{CH}_2\text{Cl})_2$ at 83 °C.

Scheme 2. Crossover Reaction between 1b and 4a

cleavage. Pleasingly, the substituted benzyl groups and 2 naphthylmethyl group in substrates 4e−g were well tolerated under the current conditions, affording 5e−g (Table 3, entries 5−7) in moderate yields.

To investigate the migration step in the current reaction, a crossover experiment was performed in which an equimolar mixture of two different alkynyl oxime ethers, 1b and 4a,was subjected to the domino reaction under the optimized conditions (Scheme 2). As a result, the noncrossover products 3b and 5a were obtained in 76 and 58% yields, respectively, together with small amounts of the crossover products 6 and 3a. This result demonstrated that the benzyl cation could partially dissociate and exchange with the other cations following the initial cyclization, as a consequence of the high reaction temperature.

Although the precise mechanism of the reaction could not be fully elucidated, a possible reaction pathway was proposed and is depicted in Scheme 3. Initially, the oxygen atom of the oxime ether adds to the Cu(II)-activated C−C triple bond in a 5-endodig fashion, generati[ng](#page-3-0) an oxonium intermediate B. The irreversible generation and 1,3-migration of the benzyl cation via the formation of ion-pair C would lead to the generation of

intermediate D. The interaction of benzyl cation and the π electron of the isoxazole core would lead to an intramolecular 1,3-migration, although partial separation of the ion-pair could occur at high temperatures.^{18a} However, an alternative pathway for the formation of crossover product, in which the intermediate B acted [as](#page-6-0) an electrophile causing the intermolecular benzylation without the generation of the cation species, would not be excluded. In the final step, aromatization of D would afford the trisubstituted isoxazole 3 and liberate the copper catalyst.

In conclusion, we have successfully developed a coppercatalyzed synthesis of trisubstituted isoxazoles from alkynyl oxime ethers via the intermolecular addition of the oxime oxygen atom to the alkyne group with subsequent 1,3 migration of the substituted benzyl group. The current synthetic method represents a typical atom-economical synthesis in that all of the atoms in starting substrate were incorporated in the final product without the loss of any atoms.

EXPERIMENTAL SECTION

General Procedure for Prepatarion of Alkynyl Oxime Ether 1a−g and 7. To a solution of alkynyl ketone (6.53 mmol) in EtOH (65 mL) were added O-(p-methoxybenzyl)hydroxylamine (1.0 g, 6.53 mmol) and PPTS (163 mg, 0.65 mmol). The reaction mixture was stirred at reflux for 12 h, then concentrated under reduced pressure, diluted with H_2O , and extracted with CHCl₃. The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on $SiO₂$ (hexane/AcOEt = 30:1−10:1) afforded the corresponding oxime ether 1a-g and 7. The geometry of C=N was deduced by NOESY experiments.

(1Z)-1,3-Diphenyl-2-propyn-1-one O-[(4-Methoxyphenyl) methyl]oxime (1a). 1,3-Diphenyl-2-propyn-1-one was used for ketone. Isolated yield 75% (1.67 g). Colorless oil. IR (neat): 2935, 2213 cm[−]¹ . 1 H NMR (300 MHz, CDCl3) δ: 7.92−7.89 (2H, m), 7.60−7.57 (2H, m), 7.43−7.36 (8H, m), 6.91 (2H, dt, J = 8.5, 2.0 Hz), 5.30 (2H, s), 3.81 (3H, s). ¹³C NMR (75 MHz, CDCl₃) δ: 159.3, 140.1, 133.7, 132.1, 129.9, 129.6, 129.6, 129.4, 128.4, 128.3, 126.5, 121.9, 113.7, 101.1, 79.7, 76.9, 55.2. HRMS (Orbitrap-ESI): calcd for $C_{23}H_{20}NO_2$ [M + H]⁺ 342.1489, found 342.1490.

(1Z)-1-[4-(Trifluoromethyl)phenyl]-3-phenyl-2-propyn-1-one O- [(4-Methoxyphenyl)methyl]oxime (1b). 3-Phenyl-1-[4-(trifluoromethyl)phenyl]-2-propyn-1-one was used for ketone. Isolated yield 41% (1.09 g). Colorless crystals. Mp: 69–71 °C (Et₂O–hexane). IR (neat): 2937, 2213 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 8.01 (2H, dd, J = 8.0, 0.5 Hz), 7.63 (2H, dd, J = 8.0, 0.5), 7.60–7.56 (2H, m), 7.42−7.37 (5H, m), 6.91 (2H, dt, J = 8.5, 2.0 Hz), 5.32 (2H, s), 3.81

(3H, s). 13C NMR (75 MHz, CDCl3) δ: 159.5, 138.9, 137.1, 132.1, 131.4, 131.0, 130.0, 129.7, 129.3, 121.5, 113.8, 101.9, 79.1, 77.3, 55.2. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{19}NO_2F_3 [M + H]^+$ 410.1362, found 410.1362.

(1Z)-1-(4-Methoxyphenyl)-3-phenyl-2-propyn-1-one O-[(4- Methoxyphenyl)methyl]oxime (1c). 1-(4-Methoxyphenyl)-3-phenyl-2-propyn-1-one was used for ketone. Isolated yield 36% (0.87 g). Colorless oil. IR (neat): 2934, 2213 cm[−]¹ . 1 H NMR (300 MHz, CDCl₃) δ : 7.84 (2H, dt, J = 8.5, 2.0), 7.58–7.55 (2H, m), 7.41–7.32 (SH, m) , 6.90 (2H, dt, J = 9.0, 2.0 Hz), 6.89 (2H, dt, J = 8.5, 2.0 Hz), 5.27 (2H, s), 3.81 (3H, s), 3.78 (3H, s). ¹³C NMR (75 MHz, CDCl₃) δ: 160.8, 159.3, 139.8, 132.1, 129.8, 129.3, 128.4, 127.9, 126.4, 122.0, 113.7, 113.7, 100.7, 79.8, 76.7, 55.3, 55.2. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{22}NO_3$ [M + H]⁺ 372.1594, found 372.1591.

(1Z)-1-Cyclohexyl-3-phenyl-2-propyn-1-one O-[(4-Methoxyphenyl)methyl]oxime (1d). 1-Cyclohexyl-3-phenyl-2-propyn-1-one was used for ketone. Isolated yield 72% (1.63 g). Colorless crystals. Mp: 78–80 °C (Et2O–hexane). IR (neat): 2930, 2212 cm^{−1}. ¹H NMR (300 MHz, CDCl₃) δ : 7.51–7.47 (2H, m), 7.35–7.29 (5H, m), 6.87 $(2H, dt, J = 8.5, 2.0 Hz)$, 5.12 $(2H, s)$, 3.78 $(3H, s)$, 2.41 $(1H, tt, J =$ 11.5, 3.5 Hz), 1.89−1.17 (10H, m). ¹³C NMR (75 MHz, CDCl₃) δ: 159.2, 146.6, 132.0, 131.9, 129.9, 129.7, 129.6, 129.1, 128.3, 122.1, 113.7, 113.6, 100.0, 79.9, 76.3, 75.9, 55.2, 42.9, 37.0, 30.7, 29.4, 25.8, 25.7. HRMS (Orbitrap-ESI): calcd for $C_{23}H_{26}NO_2$ $[M + H]^+$ 348.1958, found 348.1958.

(2E)-2-[[(4-Methoxyphenyl)methyloxy]imino]-4-phenyl-3-butynoic Acid Methyl Ester (1e). Methyl 2-oxo-4-phenyl-3-butynoate was used for ketone. Isolated yield 77% (1.62 g). Colorless oil. IR (neat): 2954, 2209, 1739 cm[−]¹ . 1 H NMR (300 MHz, CDCl3) δ: 7.53−7.50 $(2H, m)$, 7.39–7.28 (5H, m), 6.89 (2H, dt, J = 8.5, 2.0 Hz), 5.36 (2H, s), 3.89 (3H, s), 3.78 (3H, s). ¹³C NMR (75 MHz, CDCl₃) δ: 161.4, 159.7, 133.7, 132.1, 130.0, 129.7, 128.3, 128.1, 121.2, 113.8, 102.7, 78.5, 77.9, 77.2, 55.1, 53.0. HRMS (Orbitrap-ESI): calcd for $C_{19}H_{18}NO_4$ [M + H]⁺ 324.1230, found 324.1232.

(1Z)-1-Phenyl-2-heptyn-1-one O-[(4-Methoxyphenyl)methyl] oxime (1f). 1-Phenyl-2-heptyn-1-one was used for ketone. Isolated yield 99% (2.07 g). Colorless oil. IR (neat): 2958, 2217 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 7.84−7.81 (2H, m), 7.39−7.33 (5H, m), 6.89 (2H, dt, J = 8.5, 2.0 Hz), 5.24 (2H, s), 3.80 (3H, s), 2.52 (2H, t, J $= 7.0$ Hz), 1.67–1.44 (4H, m,), 0.92 (3H, t, J = 7.0 Hz). ¹³C NMR (75 MHz, CDCl₃) δ : 159.3, 140.4, 134.0, 129.9, 129.7, 129.3, 128.2, 126.4, 113.6, 103.8, 76.6, 71.7, 55.1, 30.3, 21.9, 19.4, 13.5. HRMS (Orbitrap-ESI): calcd for $C_{21}H_{24}NO_2$ [M + H]⁺ 322.1802, found 322.1804.

(1Z)-3-Cyclohexyl-1-phenyl-2-propyn-1-one O-[(4-Methoxyphenyl)methyl]oxime (1g). 3-Cyclohexyl-1-phenyl-2-propyn-1-one was used for ketone. Isolated yield 50% (1.13 g). Colorless oil. IR (neat): 2932, 2215 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 7.85–7.81 $(2H, m)$, 7.39–7.33 (5H, m), 6.89 (2H, dt, J = 8.5, 2.0 Hz), 5.24 (2H, s), 3.81 (3H, s), 2.73 (1H, m), 1.89−1.33 (10H, m). ¹³C NMR (75 MHz, CDCl₃) δ: 159.2, 140.5, 134.1, 129.7, 129.3, 128.1, 126.4, 113.6,

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107.5, 76.5, 71.7, 55.1, 32.0, 29.7, 25.7, 24.4. HRMS (Orbitrap-ESI): calcd for $C_{22}H_{26}NO_2$ [M + H]⁺ 348.1958, found 348.1948.

(1Z)-3-(Trimethylsilyl)-1-phenyl-2-propyn-1-one O-[(4-Methoxyphenyl)methyl]oxime (7). 1-Phenyl-3-trimethylsilyl-2-propyn-1-one was used for ketone. Isolated yield 68% (1.49 g). Colorless oil. IR (neat): 2959, 2157, 1614. cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 7.86−7.82 (2H, m), 7.42−7.36 (5H, m), 6.91 (2H, d, J = 8.5 Hz), 5.28 $(2H, s)$, 3.82 $(3H, s)$, 0.31 $(9H, s)$. ¹³C NMR (75 MHz, CDCl₃) δ : 159.5, 140.3, 133.6, 130.0, 129.85, 129.75, 128.6, 128.5, 126.7, 113.9, 108.4, 94.4, 77.7, 55.4, −0.1. HRMS (Orbitrap-ESI): calcd for $C_{20}H_{24}NO_2Si$ [M + H]⁺ 338.1571, found 338.1571.

Prepatarion of Alkynyl Oxime Ether 1h,i. (1E)-1-Phenyl-2propyn-1-one O-[(4-Methoxyphenyl)methyl]oxime (1h). To a solution of 7 (525 mg, 1.56 mmol) in THF (50 mL) was added dropwise TBAF $(3.1 \text{ mL}, 3.11 \text{ mmol}, 1.0 \text{ M} \text{ in } THF)$ under $N₂$ atmosphere at 0 °C. After being stirred at the same temperature for 30 min, the reaction mixture was diluted with H_2O and extracted with CHCl₃. The combined organic layers were dried over $MgSO₄$ and concentrated under reduced pressure. Purification by FCC (hexane/ AcOEt = 30:1) afforded 1h (413 mg, 99%). Colorless crystals. Mp: 72−74 °C (Et₂O-hexane). IR (neat): 2938, 2346, 1615 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 7.85−7.82 (2H, m), 7.39−7.34 (5H, m), 6.88 (2H, d, J = 8.5 Hz), 5.26 (2H, s), 3.79 (3H, s), 3.73 (1H, s). ¹³C NMR (75 MHz, CDCl₃) δ: 159.4, 139.3, 133.0, 130.0, 129.7, 129.3, 128.3, 126.3, 113.7, 89.3, 77.1, 73.6, 55.2. HRMS (Orbitrap-ESI): calcd for $C_{17}H_{16}NO_2$ [M + H]⁺ 266.1176, found 266.1176.

(1Z)-4-Hydroxy-1-phenyl-2-butyn-1-one O-[(4-Methoxyphenyl) methyl]oxime (8) . To a solution of 1h $(200 \text{ mg}, 0.75 \text{ mmol})$ in THF (11 mL) was added dropwise n-BuLi (0.71 mL, 1.13 mmol, 1.6 M in hexane) under Ar atmosphere at −40 °C. After the reaction mixture was stirred for 5 min, paraformaldehyde (33 mg, 1.13 mmol) was added. The reaction mixture was stirred for 15 min and then allowed to warm to room temperature. The reaction was quenched with satd NH4Cl and extracted with AcOEt. The combined organic layers were washed with brine, dried over $MgSO₄$, and concentrated under reduced pressure. Purification by PTLC (hexane/AcOEt = $5:1$) afforded 8 (194 mg, 88%).

(1Z)-4-[[(1,1-Dimethylethyl)dimethylsilyl]oxy]-1-phenyl-2-butyn-1-one O-[(4-Methoxyphenyl)methyl]oxime (1i). To a solution of 8 (50 mg, 0.17 mmol) in CH_2Cl_2 (2 mL) were added TBSCl (31 mg, 0.20 mmol) and imidazole (25 mg, 0.37 mmol) at room temperature. After being stirred for 2 h under N_2 atmosphere at room temperature, the reaction mixture was diluted with H₂O and extracted with CHCl₃. The organic layers dried over $MgSO₄$ and concentrated in vacuo. Purification by PTLC (hexane/AcOEt = $10:1$) afforded 1i (62 mg, 89%). Colorless oil. IR (neat): 2931, 2215 1613 cm^{−1}. ¹H NMR (300 MHz, CDCl3) δ: 7.84−7.80 (2H, m), 7.38−7.33 (5H, m), 6.88 (2H, dt, $J = 8.5, 2.0$ Hz), 5.23 (2H, s), 4.60 (2H, d, $J = 1.0$ Hz), 3.80 (3H, s), 0.91 (9H, s), 0.12 (6H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 159.4, 139.6, 133.4, 130.1, 129.5, 128.2, 126.4, 113.7, 100.4 76.9, 75.4, 55.2, 52.1, 25.7, 18.2, -5.2. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{32}NO_3Si$ $[M + H]$ ⁺ 410.2146, found 410.2145.

General Procedure for Cu(OTf)₂-Catalyzed Synthesis of Trisubstituted Isoxazoles (3a−h). To a solution of alkynyl oxime ether (30 mg) in chlorobenzene (15 mL) was added $Cu(OTf)_{2}$ (5 mol %) under an Ar atmosphere at room temperature. The reaction mixture was stirred at reflux for 2.5 h and then concentrated under reduced pressure. Purification by PTLC (hexane/EtOAc) afforded the corresponding trisubstituted isoxazole 3a−h.

3,5-Diphenyl-4-[(4-methoxyphenyl)methyl]isoxazole (3a). Isolated yield 70% (21.0 mg). Colorless solid. Mp: 98-101 °C (Et2O− hexane). IR (neat): 2931, 1610 cm^{−1}. ¹H NMR (300 MHz, CDCl₃) δ: 7.70−7.67 (2H, m), 7.55−7.52 (2H, m), 7.44−7.37 (6H, m), 7.09 $(2H, dt, J = 8.5, 2.0 Hz)$, 6.86 $(2H, dt, J = 8.5, 2.0 Hz)$, 4.04 $(2H, s)$, 3.79 (3H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 167.0, 164.2, 158.3, 130.8, 129.8, 129.5, 129.3, 128.9, 128.8, 128.7, 128.3, 128.1, 126.8, 114.3, 111.2, 55.2, 27.9. HRMS (Orbitrap-ESI): calcd for $C_{23}H_{20}NO_2$ $[M + H]$ ⁺ 342.1489, found 342.1489.

3-[4-(Trifluoromethyl)phenyl]-4-[(4-methoxyphenyl)methyl]-5 phenylisoxazole (3b). Isolated yield 85% (25.5 mg). Colorless oil. IR

(neat): 2935, 1613 cm[−]¹ ; 1 H NMR (300 MHz, CDCl3) δ: 7.71−7.68 $(2H, m)$, 7.64 $(4H, s)$, 7.45−7.43 $(3H, m)$, 7.08 $(2H, d, J = 8.5 Hz)$, 6.87 (2H, dt, J = 8.5, 2.0 Hz), 4.04 (2H, s), 3.80 (3H, s). ¹³C NMR (75 MHz, CDCl3) δ: 167.7, 163.0, 158.5, 132.9, 130.3, 130.1, 129.0, 128.8, 128.7, 127.8, 126.9, 125.7, 125.6, 114.4, 111.2, 55.3, 27.9. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{19}NO_2F_3 [M + H]^+$ 410.1362, found 410.1353.

3-(4-Methoxyphenyl)-4-[(4-methoxyphenyl)methyl]-5-phenylisoxazole (3c). Isolated yield 78% (23.4 mg). Colorless oil. IR (neat): 2935, 1612 cm[−]¹ . 1 H NMR (300 MHz, CDCl3) δ: 7.69−7.65 (2H, m), 7.47 (2H, dt, $J = 9.0$, 2.5 Hz) 7.43–7.39 (3H, m), 7.10 (2H, dt, $J = 9.0$, 2.5 Hz) 6.89 (2H, dt, $J = 9.0$, 2.5 Hz), 6.86 (2H, dt, $J = 9.0$, 2.5 Hz), 4.02 (2H, s), 3.80 (3H, s), 3.79 (3H, s). 13C NMR (75 MHz, CDCl3) δ: 166.9, 163.8, 160.6, 158.3, 130.8, 129.7, 129.6, 128.9, 128.8, 128.2, 126.8, 121.6, 114.3, 114.2, 111.0, 55.24, 55.21, 28.0. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{22}NO_3 [M + H]^+$ 372.1594, found 372.1591.

3-Cyclohexyl-4-[(4-methoxyphenyl)methyl]-5-phenylisoxazole (3d). Isolated yield 73% (21.9 mg). Colorless crystals. Mp: 128−130 °C (EtOAc–hexane). IR (neat): 2927, 1615 cm⁻¹. ¹H NMR (300 MHz, CDCl3) δ: 7.64−7.60 (2H, m), 7.41−7.37 (3H, m), 7.05 (2H, dt, $J = 8.5, 2.0$ Hz), 6.83 (2H, dt, $J = 8.5, 2.0$ Hz), 3.93 (2H, s), 3.78 $(3H, s)$, 2.49 (1H, tt, J = 13.5, 3.0 Hz), 1.86–1.52 (7H, m), 1.29–1.21 (3H, m). 13C NMR (75 MHz, CDCl3) δ: 168.5, 165.5, 158.3, 130.7, 129.4, 128.8, 128.4, 126.8, 114.1, 111.2, 55.2, 35.6, 31.6, 27.5, 26.3, 25.9; HRMS (Orbitrap-ESI): calcd for $C_{23}H_{26}NO_2$ [M + H]⁺ 348.1958, found 348.1957.

4-[(4-Methoxyphenyl)methyl]-5-phenyl-3-isoxazolecarboxylic Acid Methyl Ester (3e). Isolated yield 70% (21.0 mg). Colorless crystals. Mp: 84–86 °C (Et₂O–hexane). IR (neat): 2953, 1740, 1613 cm[−]¹ . 1 H NMR (300 MHz, CDCl3) δ: 7.69−7.66 (2H, m), 7.47−7.44 $(3H, m)$, 7.05 $(2H, d, J = 8.5 Hz)$, 6.83 $(2H, dt, J = 8.5, 2.0 Hz)$, 4.21 (2H, s), 3.91 (3H, s), 3.77 (3H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 168.6, 160.8, 158.2, 155.4, 130.6, 130.3, 129.0, 128.8, 127.2, 127.1, 114.4, 114.0, 55.2, 52.6, 27.2. HRMS (Orbitrap-ESI): calcd for $C_{19}H_{18}NO_4$ [M + H]⁺ 324.1230, found 324.1227.

5-Butyl-4-[(4-methoxyphenyl)methyl]-3-phenylisoxazole (3f). Isolated yield 66% (19.8 mg). Colorless oil. IR (neat): 2957, 1613 cm^{−1}.
¹H NMB (300 MHz, CDCl.) δ: 7.52–7.49 (2H, m), 7.38–7.36 (3H ¹H NMR (300 MHz, CDCl₃) δ: 7.52–7.49 (2H, m), 7.38–7.36 (3H, m), 6.98 (2H, dt, J = 8.5, 2.0 Hz), 6.80 (2H, dt, J = 8.5, 2.0 Hz), 3.79 $(2H, s)$, 3.77 $(3H, s)$, 2.67 $(2H, t, J = 7.5 Hz)$, 1.65 $(2H, quint, J = 7.5 Hz)$ Hz), 1.35 (2H, m), 0.90 (3H, t, J = 7.5 Hz). 13C NMR (75 MHz, CDCl3) δ: 170.9, 162.8, 158.1, 131.2, 129.7, 129.2, 128.8, 128.6, 128.2, 114.0, 111.2, 55.2, 29.5, 27.2, 25.5, 22.3, 13.6. HRMS (Orbitrap-ESI): calcd for $C_{21}H_{24}NO_2$ [M + H]⁺ 322.1802, found 322.1800.

5-Cyclohexyl-4-[(4-methoxyphenyl)methyl]-3-phenylisoxazole (3g). Isolated yield 93% (27.9 mg). Colorless crystals. Mp: 134−136 °C (EtOAc–hexane). IR (neat): 2926, 1615 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 7.49−7.46 (2H, m), 7.38−7.34 (3H, m), 6.98 (2H, dt, $J = 8.5, 2.0$ Hz), 6.80 (2H, dt, $J = 8.5, 2.0$ Hz), 3.80 (2H, s), 3.77 (3H, s), 2.71 (1H, tt, J = 11.5, 3.0 Hz), 1.83–1.64 (7H, m), 1.30–1.26 (3H, m). ¹³C NMR (75 MHz, CDCl₃) δ 174.2, 162.8, 158.1, 131.4, 129.7, 129.1, 128.8, 128.5, 128.2, 114.0, 109.8, 55.2, 36.2, 30.8, 27.1, 26.1, 25.6. HRMS (Orbitrap-ESI): calcd for $C_{23}H_{26}NO_2$ [M + H]⁺ 348.1958, found 348.1949.

4-[(4-Methoxyphenyl)methyl]-3-phenylisoxazole (3h). Isolated yield 80% (24.0 mg). Colorless oil. IR (neat): 2935, 1611 cm^{−1}. ¹H NMR (300 MHz, CDCl₃) δ : 8.06 (1H, m), 7.63–7.60 (2H, m), 7.45– 7.42 (3H, m), 7.09 (2H, dt, $J = 8.5$, 2.0 Hz), 6.84 (2H, dt, $J = 8.5$, 2.0 Hz), 3.83 (2H, s), 3.79 (3H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 161.3, 158.3, 157.2, 130.7, 129.5, 129.4, 129.0, 128.7, 128.2, 118.2, 114.1, 55.2, 28.0. HRMS (Orbitrap-ESI): calcd for $C_{17}H_{16}NO_2$ $[M + H]^+$ 266.1176, found 266.1176.

4-[(4-Methoxyphenyl)methyl]-3-phenyl-5-isoxazolemethanol (3i). Cu(OTf)₂ (4 mg, 0.01 mmol) was added to a solution of alkynyl oxime ether 1i (100 mg, 0.24 mmol) in chlorobenzene (41 mL) under an Ar atmosphere at room temperature. The reaction mixture was stirred at reflux for 2 h and then filtered. The filter cake was then washed with THF, and the filtrates were combined. A solution of TBAF (0.24 mL, 0.24 mmol, 1 M in THF) was then added to the mixture in a dropwise manner under a N_2 atmosphere at 0 °C.

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Following a 30 min period of stirring at the same temperature, the reaction mixture was diluted with H₂O and extracted with CHCl₂. The combined organic extracts were dried over $MgSO₄$ and concentrated under reduced pressure. Purification by PTLC (toluene/EtOAc = 4:1) to give 3i (46 mg, 65%). Colorless crystals. Mp: 73–75 °C (Et₂O– hexane). IR (neat): 3380, 2933, 1612 cm^{−1}. ¹H NMR (300 MHz, CDCl₃) δ : 7.50–7.47 (2H, m), 7.41–7.37 (3H, m), 6.99 (2H, dt, J = 8.5, 2.0 Hz), 6.78 (2H, dt, $J = 8.5$, 2.0 Hz), 4.62 (2H, s), 3.85 (2H, s), 3.75 (3H, s), 2.75 (1H, br s). ¹³C NMR (75 MHz, CDCl₃) δ : 168.0, 163.1, 158.2, 130.8, 129.5, 129.0, 128.7, 128.3, 114.1, 113.8, 55.2, 54.9, 27.1; HRMS (Orbitrap-ESI): calcd for $C_{18}H_{18}NO_3$ $[M + H]^+$ 296.1281, found 296.1275.

General Procedure for Preparation of Alkynyl Oxime Ethers 4a,c−g. To a solution of 1,3-diphenyl-2-propyn-1-one (1.0 g, 4.85 mmol) in EtOH (50 mL) were added alkoxyamine (4.85 mmol) and PPTS (121 mg, 0.49 mmol). The reaction mixture was stirred at reflux for 12 h, then concentrated under reduced pressure, diluted with $H₂O$ and extracted with CHCl₃. The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on SiO₂ (hexane/AcOEt = $30:1-10:1$) afforded the corresponding oxime ethers 4a,c−g.

(1Z)-1,3-Diphenyl-2-propyn-1-one O-[[4-[[(1,1-Dimethylethyl) dimethylsilyl]oxy]phenyl]methyl]oxime $(4a)$. O- $[4-[[(1,1-d)]$ Dimethylethyl)dimethylsilyl]oxy]phenyl]methyl]hydroxylamine was used for alkoxyamine. Isolated yield 71% (1.51 g). Colorless oil. IR (neat): 2930, 2214, 1610 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 8.01– 7.98 (2H, m), 7.66−7.63 (2H, m), 7.47−7.39 (8H, m), 6.92 (2H, dt, J $= 8.5, 2.0 \text{ Hz}$), 5.38 (2H, s), 1.06 (9H, s), 0.28 (6H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 155.4, 140.1, 133.7, 132.1, 130.3, 129.7, 129.6, 129.4, 128.4, 128.3, 126.5, 121.9, 119.9, 101.1, 79.7, 76.9, 25.6, 18.1, −4.5. HRMS (Orbitrap-ESI): calcd for $C_{28}H_{32}NO_2Si$ $[M + H]^+$ 442.2197, found 442.2198.

(1Z)-1,3-Diphenyl-2-propyn-1-one O-[(2-Methoxyphenyl) methyl]oxime (4c). O-[(2-Methoxyphenyl)methyl]hydroxylamine was used for alkoxyamine. Isolated yield 7% (0.115 g). Colorless oil. IR (neat): 2939, 2214, 1604 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 7.94−7.90 (2H, m), 7.62−7.59 (2H, m), 7.49 (1H, dd, J = 8.0, 2.0 Hz), 7.41−7.35 (6H, m), 7.28 (1H, td, J = 8.0, 2.0 Hz), 6.97 (1H, td, J $= 8.0, 2.0 \text{ Hz}$, 6.89 (1H, d, J = 8.0 Hz), 5.46 (2H, s), 3.86 (3H, s). ¹³C NMR (75 MHz, CDCl3) δ: 157.0, 140.2, 133.7, 132.1, 129.5, 129.4, 128.9, 128.7, 128.4, 128.3, 126.5, 126.3, 122.0, 120.4, 110.3, 101.2, 79.8, 72.1, 55.4. HRMS (Orbitrap-ESI): calcd for $C_{23}H_{20}NO_2$ [M + H]+ 342.1489, found 342.1487.

4-[[(Z)-(1,3-Diphenyl-2-propyn-1-ylidene)amino]oxymethyl] benzoic Acid Methyl Ester (4d). Methyl 4-[(aminooxy)methyl]benzoate was used for alkoxyamine. Isolated yield 32% (0.572 g). Colorless crystals. Mp: 64−68 °C (AcOEt−hexane). IR (neat): 2951, 2215, 1716, 1615 cm[−]¹ . 1 H NMR (300 MHz, CDCl3) δ: 8.04 (2H, dt, J = 8.5, 2.0 Hz) 7.91−7.88 (2H, m), 7.62−7.59 (2H, m), 7.52 (2H, d, J $= 8.5$ Hz), 7.40–7.36 (6H, m), 5.41 (2H, s), 3.90 (3H, s). ¹³C NMR (75 MHz, CDCl3) δ: 166.9, 142.9, 140.9, 133.3, 132.1, 129.8, 129.7, 129.6, 129.5, 128.5, 128.4, 127.5, 126.5, 121.7, 101.6, 79.4, 76.2, 52.0. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{20}NO_3$ $[M + H]^+$ 370.1438, found 370.1437.

(1Z)-1,3-Diphenyl-2-propyn-1-one O-[1-(4-Methoxyphenyl) ethyl]oxime (4e). O-[1-(4-Methoxyphenyl)ethyl]hydroxylamine was used for alkoxyamine. Isolated yield 39% (0.671 g). Colorless oil. IR (neat): 2939, 2207, 1613 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 7.89– 7.86 (2H, m), 7.62−7.58 (2H, m), 7.40−7.35 (8H, m), 6.89 (2H, dt, J $= 8.5, 2.0$ Hz), 5.44 (1H, q, J = 6.5 Hz), 3.79 (3H, s), 1.68 (3H, d, J = 6.5 Hz). ¹³C NMR (75 MHz, CDCl₃) δ: 159.0, 139.9, 135.2, 133.9, 132.1, 129.5, 129.4, 128.4, 128.3, 127.6, 126.5, 122.2, 113.7, 100.9, 82.0, 77.2, 55.2, 22.0. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{22}NO_2$ [M + H]+ 356.1645, found 356.1643.

(1Z)-1,3-Diphenyl-2-propyn-1-one O-(Diphenylmethyl)oxime (4f). O-(Diphenylmethyl)hydroxylamine was used for alkoxyamine. Isolated yield 64% (1.20 g). Colorless crystals. Mp: 135−138 °C (AcOEt/hexane). IR (neat): 3062, 2213, 1599 cm⁻¹. ¹H NMR (300 MHz, CDCl3) δ: 7.88−7.85 (2H, m), 7.62−7.59 (2H, m), 7.47 (4H, d, J = 8.0 Hz), 7.37−7.22 (12H, m), 6.46 (1H, s). 13C NMR (75 MHz,

CDCl3) δ: 141.4, 141.0, 133.4, 132.1, 129.7, 129.5, 128.5, 128.3, 127.6, 127.3, 126.6, 122.0, 101.5, 87.7, 80.0, 77.2. HRMS (Orbitrap-ESI): calcd for $C_{28}H_{22}NO [M + H]^+$ 388.1696, found: 388.1692.

(1Z)-1,3-Diphenyl-2-propyn-1-one O-(2-naphthylmethyl)oxime (4g). O-(2-Naphthalenylmethyl)hydroxylamine was used for alkoxyamine. Isolated yield 60% (1.05 g). Colorless solid. IR (neat): 3061, 2210, 1601 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 7.93–7.90 (3H, m), 7.87−7.82 (3H, m), 7.63−7.59 (3H, m), 7.49−7.46 (2H, m), 7.41− 7.37 (6H, m), 5.54 (2H, s). 13C NMR (75 MHz, CDCl3) δ: 140.6, 135.1, 133.7, 133.3, 133.0, 129.7, 129.5, 128.7, 128.42, 128.37, 128.1, 128.0, 127.7, 127.1, 127.0, 126.5, 126.03, 126.01, 125.9, 121.9, 101.4, 79.7, 77.2. HRMS (Orbitrap-ESI): calcd for $C_{26}H_{20}NO [M + H]$ ⁺ 362.1539, found 362.1538.

Preparation of (1Z)-1,3-Diphenyl-2-propyn-1-one O-[[4- (Dimethylamino)phenyl]methyl]oxime (4b). To a solution of O- [(4-dimethylaminophenyl)methyl]hydroxylamine hydrochloride (0.84 mmol) in MeOH (6 mL) were added 1,3-diphenyl-2-propyn-1-one (173 mg, 0.84 mmol), $Na₂SO₄$ (239 mg, 1.68 mmol), and pyridine (0.6 mL) under N₂ atmosphere at room temperature. After being stirred at the same temperature for 6 h, the reaction mixture was diluted with H_2O and extracted with CHCl₃. The organic phase was washed with H_2O , dried over $MgSO_4$, and concentrated under reduced pressure. Purification by flash column chromatography on $SiO₂$ $(Kear)$ (hexane/AcOEt = 15:1) afforded 4b (98 mg, 34%). Colorless solid. Mp: 83–87 °C (Et₂O–hexane). IR (neat): 2927, 2211, 1615 cm^{−1}. ¹H NMR (300 MHz, CDCl₃) δ: 7.92–7.89 (2H, m), 7.60–7.56 (2H, m), 7.40−7.36 (8H, m), 6.74 (2H, dt, J = 9.0, 2.0 Hz), 5.27 (2H, s), 2.95 (6H, s). 13C NMR (75 MHz, CDCl3) δ: 150.4, 139.7, 133.9, 132.1, 129.9, 129.4, 129.3, 128.3, 128.3, 126.4, 125.1, 122.0, 112.2, 100.9, 79.9, 77.4, 40.5. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{23}N_{2}O$ [M + H]⁺ 355.1805, found 355.1806.

General Procedure for $Cu(OTf)₂-Catalyzed$ Synthesis of Trisubstituted Isoxazoles (5a−c,e−g). To a solution of alkynyl oxime ether (30 mg) in chlorobenzene or dichloroethane (15 mL) was added $Cu(OTf)_{2}$ (5 mol %) under an Ar atmosphere at room temperature. The reaction mixture was stirred at reflux for 2−10 h and then concentrated under reduced pressure. Purification by PTLC (hexane/EtOAc) afforded the corresponding trisubstituted isoxazole 5a−c,e−g.

4-[4-[[(1,1-Dimethylethyl)dimethylsilyloxy]phenyl]methyl]-3,5-diphenylisoxazole (5a). Isolated yield 87% (26.1 mg). Colorless crystals. Mp: 98-101 °C (Et₂O-hexane). IR (neat) 2931, 1609 cm[−]¹ . 1 H NMR (300 MHz, CDCl3) δ: 7.70−7.67 (2H, m), 7.53−7.49 $(2H, m)$, 7.44−7.33 (6H, m), 7.01 (2H, d, J = 8.0 Hz), 6.78 (2H, dt, J $= 8.0, 2.5$ Hz), 4.02 (2H, s), 0.97 (9H, s), 0.19 (6H, s). ¹³C NMR (75 MHz, CDCl₃) δ: 167.0, 164.2, 154.2, 131.5, 129.8, 129.5, 129.3, 128.9, 128.8, 128.6, 128.4, 128.1, 126.9, 120.4, 111.4, 28.1, 25.7, 18.2, −4.4. HRMS (Orbitrap-ESI): calcd for $C_{28}H_{32}NO_2Si$ $[M + H]^+$ 442.2197, found 442.2194.

4-[[4-(Dimethylamino)phenyl]methyl]-3,5-diphenylisoxazole (5b). Isolated yield 83% (24.9 mg). Colorless crystals. Mp: 118−120 $^{\circ}$ C (Et₂O–hexane). IR (neat): 3008, 1616 cm⁻¹. ¹H NMR (300 MHz, CDCl3) δ 7.73−7.69 (2H, m), 7.58−7.55 (2H, m), 7.43−7.36 (6H, m), 7.05 (2H, d, J = 8.5), 6.70 (2H, dt, J = 8.5, 2.0 Hz), 4.00 (2H, s), 2.93 (6H, s). 13C NMR (75 MHz, CDCl3) δ: 166.9, 164.2, 149.3, 129.6, 129.4, 129.3, 128.9, 128.6, 128.4, 128.3, 128.1, 126.8, 126.5, 113.0, 111.5, 40.6, 27.7. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{23}N_{2}O$ $[M + H]$ ⁺ 355.1805, found 355.1798.

4-[(2-Methoxyphenyl)methyl]-3,5-diphenylisoxazole (5c). Isolated yield 57% (17.1 mg). Colorless crystals. Mp: 152−155 °C (EtOAc− hexane). IR (neat) 3011, 1600 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 7.67−7.64 (2H, m), 7.54−7.52 (2H, m), 7.42−7.32 (6H, m), 7.27− 7.21 (1H, m), 6.99 (1H, d, J = 7.5 Hz), 6.91 (1H, d, J = 7.5 Hz), 6.85 $(1H, t, J = 7.5 Hz)$, 4.02 $(2H, s)$, 3.82 $(3H, s)$. ¹³C NMR (75 MHz, CDCl3) δ: 167.1, 164.3, 157.1, 129.6, 129.4, 128.8, 128.6, 128.4, 128.3, 127.7, 127.1, 126.8, 120.7, 110.8, 110.0, 55.2, 23.3. HRMS (Orbitrap-ESI): calcd for $C_{23}H_{20}NO_2$ [M + H]⁺ 342.1489, found 342.1504.

4-[1-(4-Methoxyphenyl)ethyl]-3,5-diphenylisoxazole (5e). Isolated yield 54% (16.2 mg). Colorless oil. IR (neat): 2973, 1612 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 7.60–7.57 (2H, m), 7.43–7.30 (8H, m),

7.13 (2H, d, $J = 8.5$ Hz), 6.84 (2H, dt, $J = 8.5$ Hz), 4.35 (1H, q, $J = 7.0$ Hz), 3.81 (3H, s), 1.42 (3H, d, J = 7.0 Hz). 13C NMR (75 MHz, CDCl3) δ: 166.5, 164.1, 158.0, 136.1, 129.9, 129.8, 129.2, 129.0, 128.6, 128.5, 128.3, 128.1, 128.0, 117.8, 113.8, 55.2, 32.1, 19.4. HRMS (Orbitrap-ESI): calcd for $C_{24}H_{22}NO_2$ $[M + H]^+$ 356.1645, found 356.1641.

3,5-Diphenyl-4-(diphenylmethyl)isoxazole (5f). Isolated yield 76% (22.8 mg). Colorless solid. IR (neat): 3062, 1594 cm[−]¹ . 1 H NMR (300 MHz, CDCl₃) δ: 7.40−7.13 (16H, m), 7.03−7.00 (4H, m), 5.69 (1H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 167.6, 164.6, 141.3, 129.6, 129.5, 129.2, 129.1, 128.3, 128.2, 128.1, 128.0, 126.6, 116.1, 45.7. HRMS (Orbitrap-ESI): calcd for $C_{28}H_{22}NO [M + H]^+$ 388.1696, found 388.1691.

3,5-Diphenyl-4-(2-naphthylmethyl)isoxazole (5g). Isolated yield 49% (14.7 mg). Colorless oil. IR (CHCl₃): 3060, 1600 cm^{−1}. ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$ δ: 7.85 (2H, d, J = 8.0 Hz), 7.74–7.68 (3H, m), 7.58−7.53 (3H, m), 7.50−7.44 (2H, m), 7.42−7.30 (7H, m), 4.26 (2H, s). ¹³C NMR (75 MHz, CDCl₃) δ : 167.4, 164.3, 136.5, 133.7, 132.3, 129.9, 129.6, 129.2, 129.0, 128.8, 128.7, 128.3, 128.0, 127.8, 127.6, 126.8, 126.5, 126.3, 126.1, 125.7, 110.5, 29.1; HRMS (Orbitrap-ESI): calcd for $C_{26}H_{20}NO [M + H]^+$ 362.1539, found 362.1541.

Crossover Reaction. To a solution of alkynyl oxime ether 1b (45 mg, 0.11 mmol) and 4a (48.5 mg, 0.11 mmol) in chlorobenzene (38 mL) was added $Cu(OTf)_{2}$ (4.1 mg, 0.01 mmol) under Ar atmosphere at room temperature. The reaction mixture was stirred for 2 h at reflux and then concentrated. Purification by PTLC hexane/EtOAc = $15/1$) afforded 3b (35 mg, 76%), 5a (30 mg, 58%), 3a (4 mg, 10%), and 6 (5 mg, 9%).

3-[4-(Trifluoromethyl)phenyl]-4-[4-[[(1,1-dimethylethyl) dimethylsilyloxy]phenyl]methyl]-5-phenylisoxazole (6). Colorless solid. IR (neat): 2931, 1609 cm^{−1}. ¹H NMR (300 MHz, CDCl₃) δ: 7.71−7.68 (2H, m), 7.63 (4H, s), 7.46−7.44 (3H, m), 7.01 (2H, d, J $= 8.5$ Hz), 6.79 (2H, dt, J = 8.5, 2.0 Hz), 4.02 (2H, s), 0.98 (9H, s), 0.19 (6H, s). 13C NMR (75 MHz, CDCl3) δ: 167.6, 163.1, 154.4, 131.0, 130.1, 129.0, 128.8, 128.7, 127.8, 126.9, 125.63, 125.58, 120.6, 111.3, 28.0, 25.6, 18.2, −4.4. HRMS (Orbitrap-ESI): calcd for $C_{29}H_{31}NO_2F_3Si$ $[M + H]^+$ 510.2071, found 510.2069.

■ ASSOCIATED CONTENT

S Supporting Information

Copies of ${}^{1}H$ and ${}^{13}C$ NMR spectra for all compounds. This material is available free of charge via the Internet at http:// pubs.acs.org.

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Notes

The auth[ors declare no competing](mailto:miyata@kobepharma-u.ac.jp) financial interest.

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■ REFERENCES

(1) Katritzky, A. R.; Ramsden, C. A.; Scriven, E. F. V.; Taylor, R. J. K.; Joule, J. Comprehensive Heterocyclic Chemistry III; Elsevier: Oxford, 2008.

(2) (a) Carlsen, L.; Dö pp, D.; Dö pp, H.; Duus, F.; Hartmann, H.; Lang-Fugmann, S.; Schulze, B.; Smalley, R. K.; Wakefield, B. J. In Houben-Weyl Methods in Organic Chemistry; Schaumann, E., Ed.; Georg Thieme Verlag: Stuttgart, 1992; Vol. E8a, p 45. (b) Sperry, J.; Wright, D. Curr. Opin. Drug Discov. Devel. 2005, 8, 723.

(3) (a) Burrows, A. D.; Frost, C. G.; Mahon, M. F.; Raithby, P. R.; Richardson, C.; Stevenson, A. J. Chem. Commun. 2010, 46, 5064. (b) Lee, Y.; Koyama, Y.; Yonekawa, M.; Tanaka, T. Macromolecules 2009, 42, 7709.

(4) Wakefield, B. J. In Science of Synthesis: Houben-Weyl Methods of Molecular Transformations; Shaumann, E., Ed.; Georg Thieme Verlag: Stuttgart, 2001; Vol. 11, p 229.

(5) Jager, V.; Colinas, P. A. In ̈ Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry Toward Heterocycles and Natural Products; Padwa, A., Ed.; Wiley: Hoboken, 2002; Vol. 59, p 361.

(6) For recent examples, see: (a) Sanders, B. C.; Friscourt, F.; Ledin, P. A.; Mbua, N. E.; Arumugam, S.; Guo, J.; Boltje, T. J.; Popik, V. V.; Boons, G.-J. J. Am. Chem. Soc. 2011, 133, 949. (b) Budzik, B. W.; Evans, K. A.; Wisnoski, D. D.; Jin, J.; Rivero, R. A.; Szewczyk, G. R.; Jayawickreme, C.; Moncol, D. L.; Yu, H. Bioorg. Med. Lett. 2010, 20, 1363.

(7) For recent study on the synthesis of disubstituted isoxazole, see:

(a) Wang, L.; Yu, X.; Feng, X.; Bao, M. Org. Lett. 2012, 14, 2418.

(b) Dadiboyena, S.; Nefzi, A. Tetrahedron Lett. 2012, 53, 2096. (c) Dissanayake, A. A.; Odom, A. L. Tetrahedron 2012, 68, 807.

(d) Burhard, J. A.; Tchitchanov, B. H.; Carreira, E. M. Angew. Chem., Int. Ed. 2011, 50, 5379.

(8) (a) Waldo, J. P.; Larock, R. C. J. Org. Chem. 2007, 72, 9643. (b) Okitsu, T.; Potewar, T. M.; Wada, A. J. Org. Chem. 2011, 76, 3438.

(9) Denmark, S. E.; Kallemeyn, J. M. J. Org. Chem. 2005, 70, 2839. (10) (a) Kumar, J. S. D.; Ho, M. M.; Leung, J. M.; Toyokuni, T. Adv.

Synth. Catal. 2002, 344, 1146. (b) Moore, J. E.; Davies, M. W.; Goodenough, K. M.; Wybrow, R. A. J.; York, M.; Johnson, C. N.; Harrity, J. P. A. Tetrahedron 2005, 61, 6707.

(11) Jackowski, O.; Lecourt, T.; Micouin, L. Org. Lett. 2011, 13, 5664.

(12) She, Z.; Niu, D.; Chen, L.; Gunawan, M. A.; Shanja, X.; Hersh, W. H.; Chen, Y. J. Org. Chem. 2012, 77, 3627.

(13) Gayon, E.; Quinonero, O.; Lemouzy, S.; Vrancken, E.; Campagne, J.-M. Org. Lett. 2011, 13, 6418.

(14) For other recently reported methods for the synthesis of trisubstituted isoxazoles, see: (a) Nishiwaki, N.; Kobiro, K.; Hirao, S.; Sawayama, J.; Saigo, K.; Ise, Y.; Nishizawa, M.; Ariga, M. Org. Biomol. Chem. 2012, 10, 1987. (b) Kawai, H.; Sugita, Y.; Tokunaga, E.; Shibata, N. Eur. J. Org. Chem. 2012, 1295. (c) Xiang, D.; Xin, X.; Liu, X.; Zhang, R.; Yang, J.; Dong, D. Org. Lett. 2012, 14, 644. (d) Hashimoto, Y.; Takada, A.; Takikawa, H.; Suzuki, K. Org. Biomol. Chem. 2012, 10, 6003.

(15) (a) Lipshutz, B. H., Yamamoto, Y., Eds. Chem. Rev. 2008, 108, 3239. (b) Kirsch, S. F. Synthesis 2008, 3183. (c) Krause, N.; Winter, C. Chem. Rev. 2011, 111, 1994. (d) Corma, C.; Leyva-Pérez, A.; Sabater, M. J. Chem. Rev. 2011, 111, 1657.

(16) (a) Manarin, F.; Roehr, J. A.; Gay, R. M.; Brandao, R.; Menezes, P. H.; Nogueira, C. W.; Zeni, G. J. Org. Chem. 2009, 74, 2153. (b) Fürstner, A.; Heilmann, E. K.; Davies, P. W. *Angew. Chem., Int. Ed.* 2007, 46, 4760. (c) Nakamura, I.; Mizushima, Y.; Yamamoto, Y. J. Am. Chem. Soc. 2005, 127, 15022. (d) Fürstner, A.; Davies, P. W. J. Am. Chem. Soc. 2005, 127, 15024.

(17) (a) Takaya, J.; Udagawa, S.; Kusama, H.; Iwasawa, N. Angew. Chem., Int. Ed. 2008, 47, 4906. (b) Li, G.; Huang, X.; Zhang, L. Angew. Chem., Int. Ed. 2008, 47, 346. (c) Nakamura, I.; Yamagishi, U.; Song, D.; Konta, S.; Yamamoto, Y. Angew. Chem., Int. Ed. 2007, 46, 2284. (d) Cariou, K.; Ronan, B.; Mignani, S.; Fensterbank, L.; Malacria, M. Angew. Chem., Int. Ed. 2007, 46, 1881. (e) Shimada, T.; Nakamura, I.; Yamamoto, Y. J. Am. Chem. Soc. 2004, 126, 10546.

(18) (a) Nakamura, I.; Sato, T.; Terada, M.; Yamamoto, Y. Org. Lett. 2008, 10, 2649. (b) Nakamura, I.; Sato, T.; Yamamoto, Y. Angew. Chem., Int. Ed. 2006, 45, 4473.

(19) (a) Cheong, J. Y.; Im, D.; Lee, M.; Lim, W.; Rhee, Y. H. J. Org. Chem. 2011, 76, 324. (b) Nakamura, I.; Chan, C. S.; Araki, T.; Terada, M.; Yamamoto, Y. Adv. Synth. Catal. 2009, 351, 1089. (c) Peng, L.; Zhang, X.; Ma, M.; Wang, J. Angew. Chem., Int. Ed. 2007, 46, 1905.

(d) Fürstner, A.; Stelzer, F.; Szillat, H. J. Am. Chem. Soc. 2001, 123, 11863.

(20) (a) Wang, Y.-M.; Kuzniewski, C. N.; Rauniyar, V.; Hoong, C.; Toste, F. D. J. Am. Chem. Soc. 2011, 133, 12972. (b) Komeyama, K.; Takahashi, K.; Takai, K. Org. Lett. 2008, 10, 5119.

(21) Istrate, F. M.; Gagosz, F. Org. Lett. 2007, 9, 3181.

(22) For the related gold-catalyzed synthesis of isoxazoles, see: Nakamura, I.; Okamoto, M.; Terada, M. Org. Lett. 2010, 12, 2453.

(23) Ueda, M.; Ikeda, Y.; Sato, A.; Ito, Y.; Kakiuchi, M.; Shono, H.; Miyoshi, T.; Naito, T.; Miyata, O. Tetrahedron 2011, 67, 4612.

(24) Those conditions were previously used in a domino reaction involving a cyclization and Claisen-type [3,3]-sigmatropic rearrangement; see: Ueda, M.; Sato, A.; Ikeda, Y.; Miyoshi, T.; Naito, T.; Miyata, O. Org. Lett. 2010, 12, 2594.